

The Characteristics of soils at the steppe of Kherlen River Basin, Mongolia

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Introduction

The changes in human activity and climatic conditions would cause the desertification in northeast Mongolia. To prevent the desertification and conserve the grassland, the knowledge of soil characteristics are essential.

In soils of arid and semi-arid regions, calcic horizon (Bk horizon) is common. It was shown that the depth of Bk horizon is related to annual precipitation (Arkley, 1963). However, the studies about the generation time of Bk horizons were reported only in the limited areas, such as a southern and the northern part United States. Although the basic data is insufficient, it is increasing importance to clarify the process and rate of pedogenic calcium carbonate accumulation from the viewpoint of a cycle of the carbon in terrestrial ecosystem

The objective of this study is 1) to clarify the characteristic of the soils at the steppe of Kherlen River Basin 2) to obtain the value of ¹³C and ¹⁴C of pedogenic calcium carbonate in Bk horizons by the AMS (Accelerator Mass Spectrometry) method and to calculate the rate of pedogenic calcium carbonate accumulation.

Materials and methods

Study area

Fig. 1 showed the soil survey and sampling site. Five soil profiles were surveyed, consisting Baganuur (BGN), Jagalthaan (JGH), KherlenBayan-Ulaan (KBU), Underhan (UDH) and Darhan (DH). Those sites are located the steppe of Kherlen river Basin, and AWS site of the RAISE project.

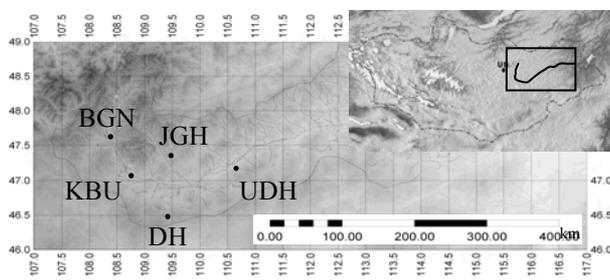


Fig. 1 Location of Soil survey and soil sampling points

Soil survey

Five soil profiles were investigated and described according to Hand book of Soil Survey (Japanese Society of Pedology, 1997). Classification of soils was followed after WRB (FAO, ISRIC, ISSS 1998). Fig. 2 showed a representative soil profile at KBU



Accumulation of calcium carbonate (calcic horizon: Bk)

Fig. 2 Soil profile of KBU

Soil sampling

Soil samples for physical measurement, chemical and isotope analysis were taken from each horizon of the five soil profiles.

Soil samples for chemical and isotope analysis were air-dried, and sieved through 2 mm. The sub samples of finely ground samples were reground to pass through 0.5 mm and 0.2 mm mesh sieve.

Undisturbed soil core samples for physical measurement were sampled by cylindrical core (100 ml) sampler.

Chemical analysis

pH(H₂O), pH(KCl) (glass electrode method), EC (platinum electrode method), Soil organic carbon and inorganic carbon contents (measured by wet combustion method (Kosaka, et al., 1959, Clark and Ogg, 1942)), Total nitrogen contents (determined by NC-analyzer using SUMIGRAPH NC-900, Sumika Chemical Analysis Service, Tokyo), Water soluble cation of Ca²⁺, Na⁺, Mg²⁺, K⁺ (determined by Atomic Absorption Spectrophotometry using AA-6200, SHIMADZU Co., Kyoto), Water soluble anions of SO₄²⁻, Cl⁻, PO₄³⁻, NO₃⁻ (determined by Ion chromatography (YOKOGA-WA IC7000 SERIES)), Exchangeable bases of Ca²⁺, Na⁺, Mg²⁺, K⁺ (determined by Atomic absorption Spectrophotometry using AA-6200, SHIMADZU Co., Kyoto) on the extract with 1 mol L⁻¹ CH₃COONH₄, CEC (measured by the method of Schollenberger (Committee of Soil Environment Analysis, 1997)

Physical measurement

Three phase ratio and saturated hydraulic conductivity were determined by the core method.

Isotopic analysis

Radio carbon and stable carbon analysis of soil carbonate samples were carried out by Accelerator Mass Spectrometry (AMS), using 5MV Tandem van de Graaff electrostatic accelerator (National Electrostatics Corporation) in Micro Analysis Laboratory, Tandem accelerator, The Univ. Tokyo.

The soil samples of Bk horizon (Fig.2) in each five profile were reacted with 85 % H₃PO₄ under vacuum to release CO₂. Then reduced to graphite with H₂ and its ¹⁴C / ¹²C ratio was measured by AMS.

$$\delta^{13}\text{C} (\text{‰}) = \left\{ \left(\frac{^{13}\text{C}/^{12}\text{C}}{\text{sample}} \right) / \left(\frac{^{13}\text{C}/^{12}\text{C}}{\text{PDB}} \right) - 1 \right\} \times 10^3$$

$$^{14}\text{C} (\text{‰}) = \left\{ \left(\frac{^{14}\text{C}/^{12}\text{C}}{\text{sample}(-25)} \right) / \left(\frac{^{14}\text{C}/^{12}\text{C}}{\text{STD}} \right) \right.$$

$$\left. \times 2^{\frac{(t - 1950)}{5730}} - 1 \right\} \times 10^3$$

$\delta^{13}\text{C}$ STD: the PDB standard, ¹⁴C STD:NIST HOX (1950),

radiocarbon age

$$= - 8033 \ln \left[\left(\frac{^{14}\text{C} / ^{12}\text{C}}{\text{sample}(-25)} \right) / \left(\frac{^{14}\text{C} / ^{12}\text{C}}{\text{STD}(-25)} \right) \right]$$

Results

The chemical and physical properties at the soil surface of the investigation area were considered to be uniform if geographical feature was a flat side. However, in the Bk horizons, difference was remarkable in pH, EC, water-soluble cation and anion, and exchangeable base especially. Water soluble ions of soils showed higher content at site with lower precipitation.

The rate of pedgenic calcium carbonate accumulation was calculated 1.10 g m⁻² yr⁻¹ from the relationships between the radiocarbon age and the contents of CaCO₃ in the Bk horizons (Fig. 3). The ¹⁴C ratio was high in the upper part in Bk horizon at the site with high amount of precipitation. In contrast, the ¹⁴C ratio was lower the upper part of Bk horizon at the site with lower amount of precipitation. The ¹³C ratio show difference of CO₂ sources, they shows difference of water movement in soils each site (Fig. 4).

Fig.5 shows the characteristics of soil at the steppe of Kherlen River Basin.

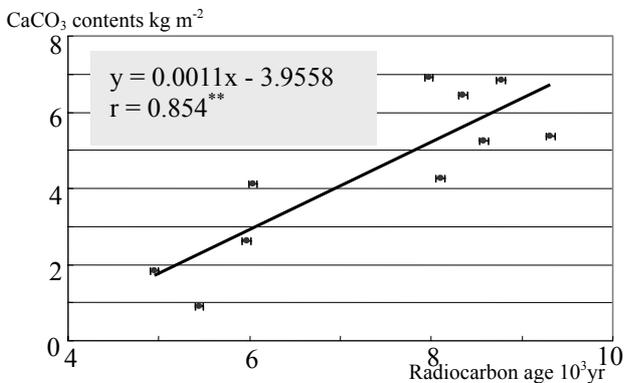


Fig. 3 The relationships between CaCO₃ contents and Radiocarbon age of pedogenic calcium carbonate in Bk horizons

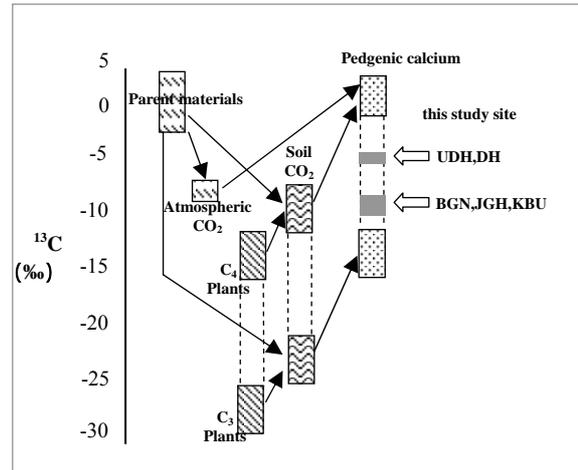


Fig. 4 The stable carbon isotope ratios of different components of terrestrial ecosystems (Nordt et al. , 1996)

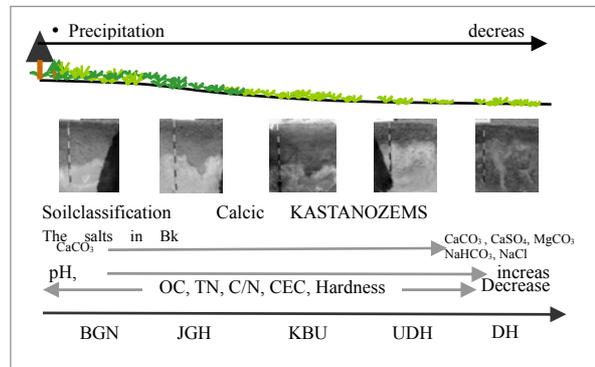


Fig. 5 The characteristics of soil at the steppe of Kherlen River Basin

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